# An Investigation of the Wittig Reaction between a Series of Monosubstituted Phthalic Anhydrides and Ethoxycarbonylmethylidenetriphenylphosphorane 

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#### Abstract

The title reaction has been investigated using a range of phthalic anhydrides substituted at the 3 - or 4 -positions with electronically dissimilar functional groups. The structures of the 3 -ethoxycarbonylmethylidenephthalides formed in these reactions have been determined by both chemical and spectroscopic means. The regioselectivity of attack by the phosphorane on an unsymmetrical phthalic anhydride appears to be largely dependent upon the electronic effects of the substituent which render one of the anhydride carbonyls relatively more or less susceptible to nucleophilic attack. In general, $(E)$-ylidenephthalides are formed predominantly.


The Wittig reaction between phthalic anhydride (1) and stabilized phosphoranes (2) was first reported in 1965, ${ }^{1}$ and shown to be a useful route to the ylidenephthalides (3) (Scheme 1). Subsequently, it has been found that ylidenebutenolides (4) and the ylidenebutanolide (5) can also be easily obtained by this method using various maleic anhydrides and succinic anhydride respectively. ${ }^{2,4}$


The former reaction has been used to prepare the highly unsaturated sesquiterpene freelingyne. ${ }^{3}$ The ylidenetetronic acid derivatives ( $6 ; \mathrm{R}^{1}=$ alkyl, aryl) can also be successfully prepared from the corresponding methoxymaleic anhydrides by this method. ${ }^{4}$ A reaction of this type constitutes the key step in a recently

(4)

(5)

(6)
reported ${ }^{5}$ synthesis of methyl multicolanate $O$-methyl ether, $(E)-\left(6 ; \quad R^{1}=n\right.$-pentyl, $\left.R^{2}=O M e\right)$, the permethylated derivative of a metabolite of Penicillium multicolor. Interestingly, reactions between phosphoranes (2; $\mathrm{R}=\mathrm{OR}^{1}$ ) and acyclic anhydrides result in the formation of acylated phosphoranes rather than the 3 -acyloxy-acrylates (7) (Scheme 2), which would be formed if the reaction proceeded in a Wittig manner as in the case of phthalic anhydride etc. ${ }^{6}$ Glutaric anhydride also undergoes an acylation rather than a Wittig reaction with phosphoranes $\left(2 ; \mathrm{R}=\mathrm{OR}^{1}\right)$, ${ }^{6}$ indicating that the latter mode of reaction only occurs in the case of anhydrides when they are cyclic and five-membered.

Similar Wittig reactions can also be carried out using 5and 6 -membered imides as substrates. ${ }^{7}$

We have recently used the reaction between 3 -methoxyphthalic anhydride and phosphorane $(2 ; \mathrm{R}=\mathrm{O}$ $\mathrm{CH}_{2} \mathrm{Ph}$ ) as the key step in a total synthesis of the fungal


Scheme 2
metabolite iso-ochracinic acid (8). ${ }^{8}$ We anticipated both on steric and electronic grounds that the phosphorane would react largely at the 7 -carbonyl of the anhydride to provide the ylidenephthalides (9) required for the synthesis of (8). In the event, the reaction was not regioselective and we obtained the ylidenephthalides ( 9 ; mixture of isomers) and (10) in a ratio of $44: 56$. Since the reasons for this lack of regioselectivity were far from clear, we decided to study the outcome of such reactions with a range of monosubstituted phthalic anhydrides. The results of this work are reported herein.

The Wittig condensations were all carried out by refluxing equivalent amounts of the anhydride and the phosphorane (2; $\mathrm{R}=\mathrm{OEt}$ ), derived from ethyl bromoacetate, in dry chloroform for 18 h . The resulting iso-
meric ylidenephthalides, which were formed in $80-85 \%$ total yield, were separated by standard chromatographic techniques. We found that the most accurate method for determining the isomeric ratios was by careful ${ }^{1} \mathrm{H}$ n.m.r. analysis of the crude reaction mixture after removal of the triphenylphosphine oxide by filtration


(10)
through silica gel eluted with ether. We found the ratios to be reproducible to $\pm 3 \%$. The results obtained are presented in the Table. The identity of each isomer was determined by a combination of ${ }^{1} \mathrm{H}$ n.m.r. data and appropriate hydrogenation experiments as follows. In the 3 -substituted phthalic anhydride series, the ( $E$ )-3-ethoxycarbonylmethylidene-7-substituted phthalides (11a) could be readily identified by the presence, in their ${ }^{1} \mathrm{H}$ n.m.r. spectra, of a double doublet ( $J 8$ and 2 Hz ), shifted downfield by approximately $\tau 1.0$ relative to the other aromatic protons, due to the proximity of the ester carbonyl group. This phenomenon has been used for assignments of structure to compounds of this type on a

Table
Products from phthalic anhydrides and phosphorane (2; $\mathrm{R}=\mathrm{OEt}$ )

(11)

(12)

$$
a ;(E)-; R^{1}=R ; R^{2}=H
$$

$$
a ;(E)-; R^{1}=R ; R^{2}=H
$$

$$
b ;(Z)-; R^{1}=R ; R^{2}=H
$$

$$
\mathrm{b} ;(Z)-\mathrm{R}^{1}=\mathrm{R} ; \mathrm{R}^{2}=\mathrm{H}
$$

$$
c ;(E)-; R^{1}=H ; R^{2}=R
$$

$$
d ;(Z)-R^{1}=H ; R^{2}=R
$$

$$
c ;(E)-R^{1}=H ; R^{2}=R
$$

$$
d ;(Z)-; R^{1}=H ; R^{2}=R
$$

\% Composition

| (11a) | (11b) | (12b) |  | R | (11c) | (11d) | (12c) | (12d) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 36 | 13 | 51 |  | OMe | 20 | t | 61 | 18 |
| 60 | 40 | $t$ ? |  | $\mathrm{NMe}_{2}$ | t | t | 65 | 35 |
| 73 | 23 | 4 |  |  | 29 | 11* | 44 | 16 * |
| 60 |  | 40 |  | $\mathrm{CO}_{2} \mathrm{Me}$ | 58 | 11 | 31 |  |
| 22 | 18 | 60 |  | $\mathrm{NO}_{2}$ | 57 | 5 | 38 |  |
| 43 | 52 | 5 |  |  | 34 | - | 46 | 20 |
| 95 | 5 | - |  | Br | 36 | 5 | 54 | 5 |

number of occasions. ${ }^{1,2,9,10}$ However, we felt that it was essential to establish the correctness of this interpretation and this we have been able to do by using $X$-ray crystallography. The compound chosen for this study was the major product from the reaction between 3-methylphthalic anhydride and phosphorane ( $2 ; \mathrm{R}=\mathrm{OEt}$ ), which was assigned structure (11aC) mainly on the basis of its ${ }^{1} \mathrm{H}$ n.m.r. spectrum [see data associated with formula (1laC)]. $X$-Ray measurements have confirmed that this compound does indeed possess the $(E)$-con-

$(11 a C)$
figuration shown. ${ }^{11}$ There are three other possible products [(11b), (12a), and (12b)] which could be formed from these types of Wittig reaction. We assumed that the other $(E)$-isomer (12a) would not be isolated due to the severe steric interaction between the substituent and the ester group; during this work, no evidence of this type of isomer was found. The two remaining $(Z)$ isomers (11b) and (12b) usually exhibited closely similar spectroscopic and chromatographic characteristics but these could be distinguished by hydrogenation to the corresponding 3 -ethoxycarbonylmethylphthalides using $10 \% \mathrm{Pd}$ on carbon in ethyl acetate. One of these products, (13), was found to be identical with the hydrogenation product arising from the ( $E$ )-isomer (11a) and was therefore derived from the $(Z)$-3-ethoxycarbonyl-methylidene- 7 -substituted phthalide (11b). Thus the remaining product was (14), derived from the $(Z)-4$ substituted phthalide (12b). In general, the hydrogenation products (13) and (14) were readily distinguished by both spectroscopic and chromatographic methods.
Much the same approach was used in the analysis of the products obtained from the 4 -substituted phthalic anhydrides. In these cases, all four possible isomers were often formed, as expected. The two ( $E$ )-isomers (11c) and (12c) were easily recognised by the occurrence in their ${ }^{1} \mathrm{H}$ n.m.r. spectra of a low-field doublet with $J 8$ or 2 Hz respectively. The $(Z)$-isomers (11d) and (12d) were then identified by a combination of the hydrogenation procedure described above, (11c) or (11d) giving the phthalides (15) while (12c) or (12d) gave phthalides (16), together with a comparison of their ${ }^{1} \mathrm{H}$ n.m.r. spectral data with those of the corresponding (E)isomers (see Experimental section).

The regioselectivity of attack by the phosphorane on the 4 -substituted phthalic anhydrides seems to be dependent largely on the electronic effects of the substituent. Electron-donating groups ( $\mathrm{NMe}_{2}, \mathrm{OMe}, \mathrm{Cl}$, Br , and Me ) cause predominant attack at the 2 -carbonyl which is meta to them. Conversely, the electron-with-



(For R, see Table)

(14)
drawing groups $\left(\mathrm{CO}_{2} \mathrm{Me}\right.$ and $\left.\mathrm{NO}_{2}\right)$ induce attack mainly at the 7 -carbonyl. While these results can be explained simply in terms of the relative electron densities of the two carbonyls (i.e. their relative electrophilicities) there is good evidence that the Wittig reactions of stabilised phosphoranes [e.g. (2)] are reversible ${ }^{12}$ and hence it may


(15)

(16)
be that the substituents affect the stabilities and rates of decomposition of the intermediate betaines. Further work is needed to establish whether or not the reactions reported herein and reversible or irreversible (e.g. by changing the nature of the substituents at phosphorus).

Much the same arguments apply in the 3 -substituted series but, in addition, steric factors, quite reasonably, play a significant role, presumably by destablising the intermediate betaines and interfering with their ability to decompose to the products. When the substituent is $\mathrm{NO}_{2}$, or $\mathrm{CO}_{2} \mathrm{Me}$, the electronic factors, however they operate, are sufficiently powerful to overcome the steric constraints.

Complexation between these substituents and the phosphorane phosphorus may help to overcome such steric factors. The total lack of regioselectivity of attack by the phosphorane on 3 -methoxyphthalic
anhydride does not fit into this pattern. By analogy with all of the other results discussed here, predominant attack at the 7 -carbonyl would be expected on both electronic and steric grounds, and it seems most likely that, in this case, complexation between the phosphorus and the oxygen of the methoxy-group plays a significant role. The coupling of 3 -methoxyphthalic anhydride with a variety of organometallic reagents has been used recently in a number of approaches to anthraquinones and anthracyclines. ${ }^{13-15}$ While many of these

(17)

(18)

(19)
reactions are almost totally regioselective, the site of attack seems to be highly dependent on the precise nature of the organometallic species and on the reaction conditions used and it is not yet possible to formulate any rules which could be used to predict the influence of electronic, steric, or complexational effects on such reactions, although it seems highly likely that the latter factor plays a role in some cases. For example, Braun ${ }^{13}$ has reported that the aryl Grignard (17) attacks only the 7 -carbonyl of 3-methoxyphthalic anhydride while Swenton's group ${ }^{14}$ has found that the closely related reagent (18) adds to this anhydride in a relatively non-regioselective manner ( $64: 23$ in favour of attack at the 2 carbonyl), and Sih et al. ${ }^{15}$ have found that the dilithiophenolate (19) attacks the 2 -carbonyl almost exclusively! A similar combination of factors is needed to explain some earlier work reported by Newman et al., in which 3-methyl- ${ }^{16}$ and 3 -chlorophthalic anhydride ${ }^{17}$ were subjected to both Grignard and Friedel-Crafts reactions. In the former anhydride, steric shielding of the 2 -carbonyl seems to be a major factor in these reactions while predictably the electronic effects are much less than in the case of 3 -chlorophthalic anhydride which undergoes exclusive attack at the 2 -carbonyl in Friedel-Crafts arylations using aluminium trichloride and either benzene or $m$-xylene. Presumably the aluminium trichloride preferentially complexes with the relatively more electron rich 2 -carbonyl.

The stereochemical outcome of the reactions reported herein is also of interest. In all cases, with the exception of 3 -chlorophthalic anhydride, the $(E)$-isomers are formed predominantly, when they are isolable (vide supra). In contrast, the classical Perkin synthesis of ylidenephthalides ( $3 ; \mathrm{R}=\mathrm{OR}^{\mathbf{1}}$ ) leads exclusively to the $(Z)$ isomers. ${ }^{18}$ This observation together with other work on the preparation of ylidenephthalides $(3 ; \mathrm{R}=\mathrm{Ph})^{\mathbf{1 0}}$ suggests that the $(Z)$-isomers of these compounds are the more thermodynamically stable. We did not observe any $(E) \longrightarrow(Z)$ isomerisation under the conditions that we have used (i.e. reflux in chloroform) to prepare the
ylidenephthalides reported here. The propensity for the formation of $(E)$-isomers in these Wittig reactions has been explained in terms of .a complexational effect between the carbonyl of the ester group and the electrondeficient aryl ring of the anhydride. ${ }^{1}$ An alternative explanation arises from a consideration of the conformations of the two possible diastereoisomers of the intermediate betaine. If it is assumed that the phosphonium

(21)
atom will occupy a position where it is flanked by two oxygen atoms then the two possibilities can be presented as projections $(20)$ and (21) which would lead to the $(E)$ (22) or ( $Z$ )-ylidenephthalides (23) respectively upon elimination of triphenylphosphine oxide. ${ }^{19}$ The diastereoisomer (20) is significantly less sterically congested and would be expected to be formed predominantly if the initial attack by the phosphorane was reversible. ${ }^{12}$

## EXPERIMENTAL

General Methods.-Melting points were determined on a Köfler hot-stage apparatus and are corrected. I.r. spectra were obtained using a Pye-Unicam SP 200G spectrometer and chloroform as solvent. U.v. spectra were measured in methanol using a Pye-Unicam SP 800 spectrometer. ${ }^{1} \mathrm{H}$ N.m.r. data were obtained using either a Perkin-Elmer R32A $(90 \mathrm{MHz})$ spectrometer or a Jeol MH-100 instrument, with $\mathrm{CDCl}_{3}$ as solvent and $\mathrm{Me}_{4} \mathrm{Si}$ as internal standard. Splittings are in Hz . Mass spectra were determined using an A.E.I. MS 902 spectrometer.

Petroleum refers to light petroleum (b.p. $40-60^{\circ} \mathrm{C}$ ) and ether to diethyl ether. Dry, ethanol-free chloroform was prepared from reagent grade material by washing the latter with equivalent volumes of water $(6 \times)$, followed by sequential drying over anhydrous magnesium sulphate and calcium chloride and finally distillation.

High-pressure liquid chromatography was carried out using stainless-steel columns packed with $5 \mu$ Partisil, in conjunction with a Waters Associates M-6000 pump and a Cecil CE 272 detector.

Phthalic Anhydrides.-The 3-nitro-, 4-nitro-, and 4-methyl-phthalic anhydrides were commercially available. 4-Chlorophthalic anhydride was obtained from commercially available 4 -chlorophthalic acid by crystallisation from hot acetic anhydride. 3-Methoxyphthalic anhydride was prepared by permanganate oxidation of 2,3 -dimethylanisole according to the method of Grewe. ${ }^{20} \quad 3$-Nitrophthalic anhydride was used as a precursor for the 3 -dimethyl-amino-, ${ }^{21} 3$-chloro, ${ }^{17}$ and 3 -bromophthalic anhydrides. ${ }^{22} 3$ Methylphthalic anhydride was obtained from the DielsAlder adduct of 2 -methylfuran and maleic anhydride using cold, concentrated sulphuric acid. ${ }^{23}$ The 3- and 4-methoxycarbonylphthalic anhydrides were prepared from the corresponding, commercial benzenetricarboxylic acids by recently described procedures. ${ }^{21}$ 4-Methoxyphthalic anhydride was prepared from methyl 3 -methoxybenzoate and chloral hydrate. ${ }^{24,21}$ 4-Dimethylaminophthalic anhydride was obtained from 4 -nitrophthalimide ${ }^{25}$ by hydrogenation in the presence of aqueous formaldehyde. ${ }^{21}$ Finally, 4bromophthalic anhydride was prepared from $\beta$-naphthol by sequential bromination and oxidation with potassium permanganate. ${ }^{22}$

Reactions between Phthalic A nhydrides and Ethoxycarbonylmethylenetriphenylphosphorane (2; $\mathrm{R}=\mathrm{OEt}$ ); General Procedure.-Equivalent amounts of the phthalic anhydride and ethoxycarbonylmethylenetriphenylphosphorane were dissolved in dry, ethanol-free chloroform ( 5 ml per 1 mmol of reactant) and the solution gently refluxed for 18 h , and then cooled and evaporated. The residue was filtered through a column of silica gel eluted with ether to remove triphenylphosphine oxide and the resulting pure mixtures of isomeric ylidenephthalides were separated as described below.

As far as possible, isomer ratios were determined by careful ${ }^{1} \mathrm{H}$ n.m.r. analysis of the product mixture after removal of triphenylphosphine oxide (see Table).

Hydrogenation of 3-Ethoxycarbonylmethylidenephthalides; General Procedure.-A solution of the phthalide ( $20-50 \mathrm{mg}$ ) in ethyl acetate ( $10-20 \mathrm{ml}$ ) containing $10 \%$ palladium on carbon catalyst (ca. 5 mg ) was shaken under hydrogen ( 1 atm ) until the appropriate volume of hydrogen had been taken up (usually $2-3 \mathrm{~h}$ ); the mixture was then filtered and the filtrate evaporated.
(E)-3-Ethoxycarbonylmethylidenephthalide (11a; $\mathrm{R}=\mathrm{H}$ ). -By the general procedure, reaction between phthalic anhydride ( 1.48 g ) and the phosphorane ( 3.48 g ) gave only one isomer, the ( E )-phthalide ( 1.90 g ), which crystallised from methanol as colourless needles, m.p. 69-70 ${ }^{\circ} \mathrm{C}$ (lit., ${ }^{2}$ m.p. $68-70^{\circ} \mathrm{C}$ ), $\lambda_{\text {max. }} 321,308,282 \mathrm{infl} ., 272,237$, and 221 nm ; $\nu_{\text {max. }} 1770,1695$, and $1640 \mathrm{~cm}^{-1} ; \tau 0.94$ (dd, $J 8$ and $2,4-\mathrm{H}), 1.91-2.31(\mathrm{~m}, 3 \mathrm{H}), 3.87\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.68(\mathrm{q}, J 7$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), and $8.66\left(\mathrm{t}, \mathrm{J} \mathrm{7}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ ) (Found: $\mathrm{C}, 66.2$; H , 4.7. $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{O}_{4}$ requires C, $66.1 ; \mathrm{H}, 4.6 \%$ ).

3-Ethoxycarbonylmethylphthalide ( $13 ; \mathrm{R}=\mathrm{H}$ ).-By the general procedure, hydrogenation of 3 -ethoxycarbonylmethylidenephthalide ( 0.05 g ) gave the phthalide ( 0.04 g ) which crystallised from petroleum as needles, m.p. 74-75 ${ }^{\circ} \mathrm{C}, \lambda_{\text {max. }} 280,272$, and 227 nm ; $\nu_{\text {max. }} 1748$ and $1712 \mathrm{~cm}^{-1}$; $\tau 2.07(\mathrm{dd}, J 8$ and $c a .2,7-\mathrm{H}), 2.19-2.57(\mathrm{~m}, 3 \mathrm{H}), 4.11(\mathrm{t}$, $J$ 7.5, $\mathrm{CHCH}_{2}$ ), 5.78 ( $\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 7.10 (d, J 7.5, $\mathrm{CHCH}_{2}$ ), and $8.75\left(\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 220(2 \%), 195(100)$,
$167(9), 146(9), 133(8)$, and $77(30)$ (Found: C, 65.4; H, 5.5. $\quad \mathrm{C}_{12} \mathrm{H}_{12} \mathrm{O}_{4}$ requires $\mathrm{C}, 65.5 ; \mathrm{H}, 5.5 \%$ ).

Reaction between 3-Methoxyphthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 3 -methoxyphthalic anhydride ( 0.14 g ) and the phosphorane $(0.27 \mathrm{~g})$ followed by silica-gel chromatography in chloroform gave the following: (i) (E)-3-ethoxycarbonylmethylidene-7methoxyphthalide (11aA) (eluted first) which crystallised from chloroform-benzene as needles, m.p. 119-120 ${ }^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 343,288$ infl., 279,27 linfl., 242 infl., and 232 nm ; $v_{\text {max }}$ 1778,1700 , and $1646 \mathrm{~cm}^{-1} ; \tau 1.30(\mathrm{~d}, J 8,4-\mathrm{H}), 2.20(\mathrm{t}$, $J 8,5-\mathrm{H}), 2.78(\mathrm{~d}, J 8,6-\mathrm{H}), 3.88\left(\cdot \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.68(\mathrm{q}, J 7$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ) $5.94(\mathrm{OMe})$, and $8.68\left(\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e$ $248(20 \%), 203(40)$, and $176(100)$ (Found: C, 62.9; H, 4.7. $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{O}_{5}$ requires $\mathrm{C}, 62.9 ; \mathrm{H}, 4.9 \%$ ); (ii) (Z)-3-ethoxycar-bonylmethylidene-4-methoxyphthalide (12bA) (eluted second) which crystallised from chloroform-petroleum as prisms, m.p. $154-155{ }^{\circ} \mathrm{C}, \lambda_{\text {max. }} 338,308,283$ infl., $273,236 \mathrm{infl}$., and 227 nm ; $\nu_{\text {max }} 1792,1710$, and $1658 \mathrm{~cm}^{-1} ; \tau 2.36-2.52$ $(\mathrm{m}, 2 \mathrm{H}), \stackrel{\text { max }}{2.76}\left(\mathrm{dd}, J 8\right.$ and 2), $3.74\left({ }^{\circ} \mathrm{CHCO} \mathrm{C}_{2} \mathrm{Et}\right), 5.71$ (q, $\left.J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 5.98(\mathrm{OMe})$, and $8.68\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e$ $248(9 \%)$, 203(36), and $176(100)$ (Found: C, 62.9; H, 4.9\%) ; and (iii) (Z)-3-ethoxycarbonylmethylidene-7-methoxyphthalide (llbA) (eluted last), which crystallised from methanol as needles m.p. $132-133{ }^{\circ} \mathrm{C}$, $\lambda_{\text {max }} 340$, 284infl., 276, $270 \mathrm{infl} .$, 244infl., and 230 nm ; $\nu_{\text {max. }} 1794,1705$, and $1668 \mathrm{~cm}^{-1}$; $\tau$ $2.28(\mathrm{t}, J 8,5-\mathrm{H}), 2.73$ (d, $J 8$ ), 2.92 (d, $J 8), 4.17$ ( $\mathrm{CHCO}_{2}{ }^{-}$ Et ), $5.70\left(\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 5.98(\mathrm{OMe})$, and $8.69(\mathrm{t}, J 7$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ); $m / e 248(14 \%), 203(53)$, and $176(100)$ (Found: C, 62.5 ; H, $4.6 \%$ ).

Hydrogenation of Products from 3-Methoxyphthalic Anhy-dride.-By the general procedure, hydrogenation of $(E)$ or (Z)-3-ethoxycarbonylmethylidene-7-methoxyphthalide (11aA) or (11bA) gave 3-ethoxycarbonylmethyl-7-methoxyphthalide (13A), a colourless oil showing $\lambda_{\text {max. }} 296$ and 234 $\mathrm{nm} ; \nu_{\text {max. }} 1758$ and $1710 \mathrm{~cm}^{-1} ; \tau 2.37(\mathrm{t}, J \mathrm{~J}, 5-\mathrm{H}), 2.96$ $(\mathrm{d}, J 8), 3.04(\mathrm{~d}, J 8), 4.20\left(\mathrm{t}, J 7.5, \mathrm{CHCH}_{2}\right), 5.79(\mathrm{q}, J 7$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.01(\mathrm{OMe}), 7.16\left(\mathrm{~d}, J 7.5, \mathrm{CHCH}_{2}\right)$, and 8.75 ( t , $J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) (Found: $M^{+}, 250.0840 . \mathrm{C}_{13} \mathrm{H}_{14} \mathrm{O}_{5}$ requires M, 250.0841).

By contrast, hydrogenation of ( $Z$ )-3-ethoxycarbonyl-methylidene-4-methoxyphthalide (12bA) gave 3 -ethoxy-carbonylmethyl-4-methoxyphthalide (14A) as an oil, with $\lambda_{\text {max. }}$. 295 and 232 nm ; $\nu_{\text {max. }} 1760$ and $1714 \mathrm{~cm}^{-1}$; $\tau 2.45-2.52$ (m, 2 H ), 2.88 (dd, J 8 and 2), 4.11 (dd, J 9 and 4, $\mathrm{CHCH}_{2}$ ), $5.83\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.09(\mathrm{OMe}), 6.71$ (dd, $J 16$ and 4, $\mathrm{CHCH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}}$ ), 7.40 (dd, $J 16$ and $9, \mathrm{CHCH}_{\mathrm{a}} H_{\mathrm{b}}$ ), and 8.79 ( t , $J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) (Found: $M^{+}, 250.0846$ ).

Reaction between 3-Dimethylaminophthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 3 -dimethylaminophthalic anhydride ( 0.63 g ) and the phosphorane ( 1.16 g ) gave a mixture of two isomers ( 0.70 g) which were separated by silica-gel chromatography in petroleum-ether ( $1: 1$ ) to give the following: (i), (E)-7-di-methylamino-3-ethoxycarbonylmethylidenephthalide (11aB) (eluted first) which crystallised from methanol as fine yellow needles, m.p. $82-83{ }^{\circ} \mathrm{C}, \lambda_{\max } 422,366$, and 276 nm ; $\nu_{\text {max }}$. 1784,1716 , and $1660 \mathrm{~cm}^{-1}$; $\tau 1.54(\mathrm{~d}, J 84-\mathrm{H}), 2.46(\mathrm{t}$, $J 8,5-\mathrm{H}), 3.03(\mathrm{~d}, J 8,6-\mathrm{H}), 4.02$ (: $\mathrm{CHCO} \mathrm{CO}_{2} \mathrm{Et}$ ), 5.76 (q, $J$ $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.93\left(\mathrm{NMe}_{2}\right)$, and $8.70\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 261$ $(91 \%), 246(33), 216(22)$, and $188(100)$ (Found: $N, 5.2$. $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{NO}_{4}$ requires $\mathrm{N}, 5.4 \%$ ); and (ii) ( $Z$ )-7-dimethylamino${ }^{9} 3$-ethoxycarbonylmethylidenephthalide (11bB) (eluted second) which crystallised from methanol as yellow prisms partially melting at $82-83{ }^{\circ} \mathrm{C}$ (thermal isomerisation?) and com-
pletely at $97-98{ }^{\circ} \mathrm{C}$ [mixed m.p. of $(E)$ - and $(Z)$-isomers, $\left.69-79^{\circ} \mathrm{C}\right], \lambda_{\text {max }} 420,368$, and 273 nm ; $\nu_{\text {max }} 1781,1714$, and $1665 \mathrm{~cm}^{-1} ; \tau 2.50(\mathrm{t}, J 8,5-\mathrm{H}), 2.96(\mathrm{~d}, J 8), 3.10(\mathrm{~d}$, $J 8), 4.23\left(:{ }^{\circ} \mathrm{CHCO}{ }_{2} \mathrm{Et}\right), 5.74\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.91\left(\mathrm{NMe}_{2}\right)$, and $8.69\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 261(48 \%), 246(35), 216(17)$, and $188(100)$ (Found: C, 64.7; H,5.5; N, 5.6. $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{NO}_{4}$ requires $\mathrm{C}, 64.4 ; \mathrm{H}, 5.7 ; \mathrm{N}, 5.4 \%$ ).

Hydrogenation of the Products from 3-Dimethylaminophthalic Anhydride.-By the general procedure, hydrogenation of either ( $E$ )-7-dimethylamino-3-ethoxycarbonylmethylidenephthalide (11aB) or its ( $Z$ )-isomer (11bB) gave 7-dimethylamino-3-ethoxycarbonylmethylphthalide (13B) as a viscous yellow oil which resisted attempts at crystallisation, and which showed $\lambda_{\text {max. }} 326$ and 252 nm ; $v_{\text {max. }} 1756$ and $1738 \mathrm{~cm}^{-1}$; $\tau 2.57(\mathrm{t}, J 8,5-\mathrm{H}), 3.20(\mathrm{~d}, J 8), 3.24(\mathrm{~d}, J 8)$, $4.30\left(\mathrm{t}, J 7.5, \mathrm{CHCH}_{2}\right), 5.82\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.92\left(\mathrm{NMe}_{2}\right)$, $7.17\left(\mathrm{~d}, J 7.5 \mathrm{CHCH}_{2}\right)$, and $8.73\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e$ 263(100\%), 248(22), 190(89), 176(60), and 172(25) (Found: $M^{+}, 263.1142 . \quad \mathrm{C}_{14} \mathrm{H}_{17} \mathrm{NO}_{4}$ requires $M, 263.1157$ ).

Reaction between 3-Methylphthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 3 -methylphthalic anhydride ( 1.62 g ) and the phosphorane $(3.48 \mathrm{~g})$ gave a mixture of two isomers $(1.98 \mathrm{~g})$ which were separated by chromatography over silica gel eluted with petroleum-ether ( $1: 1$ ) to give the following: (i) (E)-3-ethoxycarbonylmethylidene-7-methylphthalide (11aC) (eluted first) which crystallised from methanol as thick, colourless needles, m.p. $78-79^{\circ} \mathrm{C}, \lambda_{\text {max. }} 331,317,285$ infl., 275 , and 241 $\mathrm{nm} ; \nu_{\text {max }} 1768,1694$, and $1637 \mathrm{~cm}^{-1} ; \tau 1.09$ (dd, $J 8$ and 2 , $4-\mathrm{H}), 2.32(\mathrm{t}, J 8,5-\mathrm{H}), 2.55(\mathrm{~d}, J 8,6-\mathrm{H}), 3.91\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right)$, $5.74\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.32\left(\mathrm{CH}_{3}\right)$, and $8.68\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; $m / e ~ 232(33 \%), 187(100), 186(50), 160(31)$, and $103(29)$ (Found: C, 67.2; H, 5.1. $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{O}_{4}$ requires $\mathrm{C}, 67.2 ; \mathrm{H}$, $5.2 \%$ ) ; and (ii), (Z)-3-ethoxycarbonylmethylidene-7-methylphthalide (11bC) which crystallised from methanol as colourless cubes, m.p. $92-93{ }^{\circ} \mathrm{C}$ (with softening at $80^{\circ} \mathrm{C}$ ), $\lambda_{\text {max }}$ $328,315,283$ infl., 273 , and 250 nm ; $\nu_{\text {max }} 1779,1690$, and $1659 \mathrm{~cm}^{-1} ; \tau 2.3-2.6(\mathrm{~m}, 3 \mathrm{H}), 4.17\left(: \mathrm{CH} \mathrm{CO}_{2} \mathrm{Et}\right), 5.70(\mathrm{q}, J$ $\left.7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.33\left(\mathrm{CH}_{3}\right)$, and $8.68\left(\mathrm{t}, \mathrm{J} \mathrm{7}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e$ $232(24 \%), 187(100), 186(41), 160(33)$, and 103(20) (Found: C, $67.3 ; \mathrm{H}, 5.2 \%$ ). A trace of a third compound, probably the $(Z)$-4-methyl isomer was also detected [ $\tau 4.09\left(: \mathrm{CHCO}_{2}-\right.$ Et) and $\left.7.41\left(\mathrm{CH}_{3}\right)\right]$.

3-Ethoxycarbonylmethyl-7-methylphthalide (13C).Hydrogenation, by the general procedure, of either the $(E)$ 7 -methylphthalide (11aC) or the ( $Z$ )-7-methylphthalide (11bC) gave the phthalide (13C) as a colourless, viscous oil, $\lambda_{\text {max. }} 286,278$, and 230 nm ; $\nu_{\text {max. }} 1750$ and $1726 \mathrm{~cm}^{-1}$; $\tau$ $2.39 — 2.83(\mathrm{~m}, 3 \mathrm{H}), 4.21\left(\mathrm{t}, J 7.5, \mathrm{CHCH}_{2}\right), 5.81(\mathrm{q}, J 7$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.17\left(\mathrm{~d}, J 7.5, \mathrm{CHCH}_{2}\right), 7.33\left(\mathrm{CH}_{3}\right)$, and $8.76(\mathrm{t}$, $J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) (Found: $M^{+}, 234.0890 . \mathrm{C}_{13} \mathrm{H}_{14} \mathrm{O}_{4}$ requires $M, 234.0892$ ).

Reaction between 3-Methoxycarbonylphthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 3 -methoxycarbonylphthalic anhydride ( 0.68 g ) and the phosphorane ( 1.16 g ) gave a mixture of two isomers $(0.78 \mathrm{~g})$ which were separated by silica-gel chromatography in petroleum-ether (3:2) to give the following: (i) ( $E$ )-3-ethoxycarbonylmethylidene-7-methoxycarbonylphthalide
(llaD) (eluted first) which crystallised from methanol as fine needles, m.p. $105-106{ }^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 324,317.5$ infl., 314,282 , 273, 248, and 226 nm ; $\nu_{\text {max. }} 1793,1720,1700$, and 1652 $\mathrm{cm}^{-1} ; \tau 1.05(\mathrm{dd}, J 8$ and $2,4-\mathrm{H}), 2.32-2.49(\mathrm{~m}, 2 \mathrm{H}), 4.04$ ( $\left.: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.82\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.07(\mathrm{OMe})$, and $8.67(\mathrm{t}$, $\left.J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 276(23 \%), 232(11), 231(100)$, and 204(41)
(Found: C, 60.8; H, 4.5. $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{O}_{6}$ requires $\mathrm{C}, 60.9$; H , $4.4 \%$ ) ; and (ii) (Z)-3-ethoxycarbonylmethylidene-4-methoxycarbonylphthalide (12bD) (eluted second) which crystallised from methanol as fine needles, m.p. $122-123{ }^{\circ} \mathrm{C}, \lambda_{\max }$. 320 infl ., 308 infl ., $283,253 \mathrm{infl}$., and 242 nm ; $v_{\text {max. }} 1800$, 1723,1699 , and $1658 \mathrm{~cm}^{-1} ; \tau 1.90(\mathrm{~d}, J 8,5-\mathrm{H}), 2.09(\mathrm{~d}$, $J 8,7-\mathrm{H}), 2.51(\mathrm{t}, J 8,6-\mathrm{H}), 3.12$ (: $\left.\mathrm{CHCO} \mathrm{C}_{2} \mathrm{Et}\right), 5.78(\mathrm{q}, J 7$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.06(\mathrm{OMe})$, and $8.67\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e$ $276(4 \%), 231(100)$, and $204(45)$ (Found: C, 60.7; H, $4.3 \%$ ).

Hydrogenation of Products from 3-Methoxycarbonylphthalic Anhydride.-Hydrogenation of (E)-3-ethoxycar-bonylmethylidene-7-methoxycarbonylphthalide (1laD) (50 mg ) by the general procedure gave 3 -ethoxycarbonylmethyl-7methoxycarbonylphthalide (13D) ( 45 mg ) as a colourless oil, showing $\lambda_{\text {max. }} 282 \mathrm{~nm}$; $\nu_{\text {max. }} 1778$ and $1734 \mathrm{~cm}^{-1}$; $\tau 2.42-$ $2.64(\mathrm{~m}, 3 \mathrm{H}), 4.30\left(\mathrm{t}, J 7.5, \mathrm{CHCH}_{2}\right), 5.83\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $6.10\left(\mathrm{CO}_{2} \mathrm{Me}\right), 7.17\left(\mathrm{~d}, J 7.5, \mathrm{CHCH}_{2}\right)$, and 8.77 ( $\mathrm{t}, J 7$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ); $m / e$ 278(14\%), 204(69), and 191(100) (Found: $M^{+}$, 278.078. $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{O}_{6}$ requires $M, 278.079$ ).
Hydrogenation of ( $Z$ )-3-ethoxycarbonylmethylidene-4methoxycarbonylphthalide (12bD) ( 28 mg ) gave 3 -ethoxy-carbonylmethyl-4-methoxycarbonylphthalide (14D) ( 26 mg ) as colourless prisms, m.p. $54-55^{\circ} \mathrm{C}$ (ether-petroleum), $\lambda_{\text {max. }}$ 290 and 281 nm ; $\nu_{\text {max }} 1775$ and $1736 \mathrm{~cm}^{-1}$; $\tau 2.06(\mathrm{~d}, J 8$, $5-\mathrm{H}), 2.27(\mathrm{~d}, J 8,7-\mathrm{H}), 2.70(\mathrm{t}, J 8,6-\mathrm{H}), 4.07(\mathrm{dd}, J 7$ and 4, $\mathrm{CHCH}_{2}$ ), $5.96\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.08\left(\mathrm{CO}_{2} \mathrm{Me}\right), 6.78$ (dd, $J 16.5$ and $4, \mathrm{CHCH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}}$ ), 7.42 (dd, $J 16.5$ and $7, \mathrm{CH} \cdot$ $\mathrm{CH}_{\mathrm{a}} H_{\mathrm{b}}$ ), and $8.89\left(\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; $m / e$ 278(2\%), 204(81), and $191(100)$ (Found: C, 60.4; H, 4.9. $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{O}_{6}$ requires $\mathrm{C}, 60.4 ; \mathrm{H}, 5.1 \%$ ). The two compounds were readily distinguished by t.l.c.; e.g. in hexane-ether (3:2), (13D) had $R_{\mathrm{F}} 0.09$ while (14D) had $R_{\mathrm{F}} 0.18$.

Reaction between 3-Nitrophthalic Anhydride and the Phos-phorane.-By the general procedure, reaction between 3nitrophthalic anhydride ( 1.93 g ) and the phosphorane ( 3.48 $\mathrm{g})$, gave a mixture of three isomers $(2.1 \mathrm{~g})$. Chromatotography of the mixture over silica gel eluted with chloroform gave the following: (i) (E)-3-ethoxycarbonylmethylidene-7-nitrophthalide (1laE) (eluted first) which crystallised from methanol as needles, m.p. $105-106{ }^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 321,304 \mathrm{infl}$., 267 infl ., and 225 nm ; $\nu_{\text {max. }} 1793,1703$, and $1650 \mathrm{~cm}^{-1}$; $\tau 0.60(\mathrm{dd}, J 8$ and $2,4-\mathrm{H}), 1.90-2.10(\mathrm{~m}, 2 \mathrm{H}), 3.73$ (: $\mathrm{CH} \mathrm{CO}_{2} \mathrm{Et}$ ), 5.76 (q, J 7, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), and 8.71 ( $\mathrm{t}, J 7, \mathrm{CH}_{2}-$ $\mathrm{CH}_{3}$ ) ; $\mathrm{m} / e 263(35 \%), 219(27)$, and $218(100)$ (Found: C, 54.7 ; $\mathrm{H}, 3.9$; $\mathrm{N}, 5.4 . \quad \mathrm{C}_{12} \mathrm{H}_{9} \mathrm{NO}_{6}$ requires $\mathrm{C}, 54.8 ; \mathrm{H}, 3.4 ; \mathrm{N}$, $5.3 \%$ ) ; (ii) (Z)-3-ethoxycarbonylmethylidene-4-nitrophthalide (12bE) (eluted last), which crystallised from methanol or ethyl acetate as prisms, m.p. $134-135{ }^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 320.5$, 286infl., 256, and 222 nm ; $\nu_{\text {max }} 1792,1709$, and $1649 \mathrm{~cm}^{-1}$; $\tau 1.58(\mathrm{~d}, J 8), 1.70(\mathrm{~d}, J 8), 2.11(\mathrm{t}, J 8), 3.38\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right)$, 5.69 ( $\mathrm{q}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), and 8.67 ( $\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) ; m/e $263(25 \%)$, 219(19), and 218(100) (Found: C, 54.6; H, 3.5; $\mathrm{N}, 5.1 \%$ ); and (iii) a mixture of (11aE) and ( $Z$ )-3-ethoxy-carbonylmethylidene-7-nitrophthalide (llbE) (eluted second) which was not further separated; (11bE) showed a characteristic resonance at $\tau 3.60\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right)$.

The ( $Z$ )-4-nitrophthalide ( 12 bE ) was also partly separated from the mixture of isomers by fractional crystallisation with ethyl acetate.

Hydrogenation of the Products from 3-Nitrophthalic Anhydride.-By the general procedure, hydrogenation of (E)-3-ethoxycarbonylmethylidene-7-nitrophthalide (1laE) or a mixture of (1laE) and its ( $Z$ )-isomer (llbE) [see (iii) above] resulted in reduction of the nitro-group as well as the
exocyclic double-bond and gave 7-amino-3-ethoxycarbonyl. methylphthalide ( $13 ; \mathrm{R}=\mathrm{NH}_{2}$ ) as a viscous oil showing $\nu_{\text {max }}$. 3370,1738 , and $1710 \mathrm{~cm}^{-1} ; \tau 2.58-2.79(\mathrm{~m}, 2 \mathrm{H}), 3.39$ (dd, $J 8$ and 2), $4.24\left(\mathrm{t}, J 7, \mathrm{CHCH}_{2}\right), 5.82\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $7.20\left(\mathrm{~d}, J 7, \mathrm{CHCH}_{2}\right)$, and 8.76 ( $\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) (Found: $M^{+}, 235.0838 . \quad \mathrm{C}_{12} \mathrm{H}_{13} \mathrm{NO}_{4}$ requires $M, 235.0845$ ).

Similar reduction of the ( $Z$ )-4-nitrophthalide ( 12 bE ) gave 4-amino-3-ethoxycarbonylmethylphthalide (14; $\mathrm{R}=\mathrm{NH}_{2}$ ) as yellow needles, m.p. $134-135{ }^{\circ} \mathrm{C}(\mathrm{EtOH}), \lambda_{\text {max. }} 392$ and 321 nm ; $\nu_{\text {max. }} 3400,1752$, and $1718 \mathrm{~cm}^{-1}$; $\tau 2.48-2.68(\mathrm{~m}$, 2 H ) $, 3.01(\mathrm{~m}, 1 \mathrm{H}), 4.06$ (apparent, $\left.J 7, \mathrm{CHCH}_{2}\right), 5.74(\mathrm{q}, J$ $7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 6.87 (dd, $J 17$ and 6, $\mathrm{CH} \cdot \mathrm{CH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}}$ ), 7.20 (dd, $J 17$ and $7, \mathrm{CHCH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}}$ ), and $8.74\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 235(42 \%)$, $161(65), 148(100)$, and $120(17)$ (Found: C, 61.7; H, 5.5; $\mathrm{N}, 5.9 . \mathrm{C}_{12} \mathrm{H}_{13} \mathrm{NO}_{4}$ requires C, $61.3 ; \mathrm{H}, 5.5 ; \mathrm{N}, 6.0 \%$ ).

Reaction between 3-Chlorophthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 3 -chlorophthalic anhydride ( 0.91 g ) and the phosphorane $(1.74 \mathrm{~g})$ gave a mixture of two isomers ( 1.1 g ) which were separated by silica-gel chromatography in petroleumether (2:3) to give the following: (i) (E)-7-chloro-3-ethoxycarbonylmethylidenephthalide (11aF) (eluted first) which crystallised from methanol as needles, m.p. $116-117^{\circ} \mathrm{C}, \lambda_{\text {max. }} 336$, 321.5, 283 infl., 275 , and 228 nm ; $v_{\text {max. }} 1790,1710$, and $1658 \mathrm{~cm}^{-1} ; \tau 1.02(\mathrm{dd}, J 8$ and $2,4-\mathrm{H}), 2.20-2.45(\mathrm{~m}, 2 \mathrm{H})$, $3.87\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.74\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, and $8.68(\mathrm{t}, J 7$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ) ; $m / e 254(4 \%)$, 252(13), 209(24), 207(100), 180(36), and $123(16)$ (Found: C, $57.0 ; \mathrm{H}, 3.6 . \quad \mathrm{C}_{12} \mathrm{H}_{9} \mathrm{ClO}_{4}$ requires C, $57.0 ; \mathrm{H}, 3.6 \%$ ) ; and (ii) (Z)-7-chloro-3-ethoxycarbonylmethylidenephthalide (11bF) which crystallised from ethanol as needles, m.p. $121-122{ }^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 323.5,292,265,256.5$, and 233 nm ; $\nu_{\text {max. }} 1800,1710$, and $1662 \mathrm{~cm}^{-1}$; $\tau 2.10(\mathrm{dd}$, $J 8$ and 2), 2.38-2.55 (m, 2 H ), $3.41\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.71(\mathrm{q}$, $\left.J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, and $8.68\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 254(2 \%)$, $252(6), 209(23), 207(100), 180(32)$, and $123(25)$ (Found: C, and $56.9 ; \mathrm{H}, 3.7 \%$ ).

A small amount of a third isomer ( $Z$ )-4-chloro-3-ethoxycarbonylmethylidenephthalide ( 12 bF ) was also detected, $\tau$ $2.30-2.36$ (bs, 3 H ), $4.10\left(: \mathrm{CHCO}_{2} \mathrm{~F} . \mathrm{t}\right), 5.68\left(\mathrm{q}, \mathrm{J} 7 \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, and $8.66\left(\mathrm{t}, \mathrm{J} \mathrm{7}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 254(6 \%)$, 252(20), 209(20), and $207(100)$.

Hydrogenation of the Products from 3-Chlorophthalic Anhydride.-By the general procedure, hydrogenation of either ( $E$ )- or ( $Z$ )-7-chloro-3-ethoxycarbonylmethylidenephthalide [(11aF) or (llbF)] gave 7-chloro-3-ethoxycarbonylmethylphthalide ( 13 F ), as colourless needles (from petroleum-ether), m.p. $160-161{ }^{\circ} \mathrm{C}, \lambda_{\max }$ 279.5, and 272 nm ; $\nu_{\text {max }} 1759$ and $1727 \mathrm{~cm}^{-1} ; \tau 2.08$ (dd, $J 8$ and 2 ), $2.28-2.58(\mathrm{~m}, 2 \mathrm{H}), 4.12\left(\mathrm{t}, J 7.5, \mathrm{CHCH}_{2}\right), 5.79(\mathrm{q}, J 7$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.12\left(\mathrm{~d}, J 7.5, \mathrm{CHCH}_{2}\right)$, and $8.74\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; $m / e ~ 256(<1 \%)$, 254(1), 182(11), 180(31), 169(7), 167(26), and $146(100)$ (Found: $M^{+}, 254.031 ; \mathrm{C}_{12} \mathrm{H}_{11}{ }^{35} \mathrm{ClO}_{4}$ requires $M, \mathbf{2 5 4 . 0 3 5}$ ) (Found: C, 56.7; H,4.0. $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{ClO}_{4}$ requires C, $56.6 ; \mathrm{H}, 4.3 \%$ ).
(E)-7-Bromo-3-ethoxycarbonylmethylidenephthalide (11aG). -By the general procedure, reaction between 3- bromophthalic anhydride ( 0.23 g ) and the phosphorane ( 0.35 g ) followed by crystallisation of the product from methanol gave the phthalide ( 11 aG ) $(0.2 \mathrm{~g})$ as prisms, m.p. $107-108^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 334.321,282$ infl., 274, and 230 nm ; $\nu_{\text {max. }} 1781,1700$, and $1640 \mathrm{~cm}^{-1} ; \tau 0.97(\mathrm{dd}, J 8$, and $2,4-\mathrm{H}), 2.1-2.4(\mathrm{~m}, 2$ H ), $3.91\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.77\left(\mathrm{q}, J 7, \mathrm{CHCH}_{3}\right)$, and $8.70(\mathrm{t}, J 7$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; $m / e 298(31 \%), 296(33), 270(15), 268(16), 253(98)$, $251(100)$, 226(57), and 224(65) (Found: C, 48.8; H, 3.1. $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{BrO}_{4}$ requires $\mathrm{C}, 48.5 ; \mathrm{H}, 3.0 \%$ ).

Reaction between 4-Methoxyphthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 4-methoxyphthalic anhydride ( 0.09 g ) and the phosphorane $(0.174 \mathrm{~g})$ gave a mixture of three isomers $(0.11 \mathrm{~g})$. Chromatography over silica gel eluted with petroleum-ether ( $1: 1$ ) gave the following: (i) (E)-3-ethoxycarbonylmethylidene-5methoxyphthalide (12cA) (eluted first) which crytallised from methanol as plates, m.p. $112-113^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 327,295,274$ infl., 265 , and 241 nm ; $\nu_{\text {max. }} 1786,1706$, and $1650 \mathrm{~cm}^{-1}$; $\tau 1.39$ (d, J $2,4-\mathrm{H}$ ), $2.17(\mathrm{~d}, J 8,7-\mathrm{H}), 2.83$ (dd, $J 8$ and $2,6-\mathrm{H}$ ), 3.91 (: CHCO$)_{2} \mathrm{Et}$ ), 5.74 (q, J 7, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $6.06(\mathrm{OMe})$, and 8.68 (t, $J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ); $m / e 248(12 \%)$, 220(9), 203(58), and 176(100) (Found: C, 62.8; H, 4.8. $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{O}_{5}$ requires C , $62.9 ; \quad \mathrm{H}, 4.9 \%$ ); (ii) (Z)-3-ethoxycarbonylmethylidene-5methoxyphthalide (12dA) (eluted last) which crystallised from methanol as thick needles, m.p. $146-147{ }^{\circ} \mathrm{C}$, $\lambda_{\text {max }}$ 320 infl., 292, 27 linfl., 263, and 224 nm ; $v_{\text {max. }} 1794,1704$, and $1665 \mathrm{~cm}^{-1}$; $\tau 2.16(\mathrm{~d}, J 8,7-\mathrm{H}), 2.80(\mathrm{dd}, J 8$ and 2 , $6-\mathrm{H}), 2.86(\mathrm{~d}, J 2,4-\mathrm{H}), 4.19\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.71\left(\mathrm{q}, J 7, \mathrm{CH}_{2}{ }^{-}\right.$ $\left.\mathrm{CH}_{3}\right), 6.08(\mathrm{OMe})$, and $8.69\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 248(12 \%)$, $220(8), 203(60)$, and $176(100)$ (Found: C, 62.7; H, 4.7\%); and (iii) a mixture of ( 12 cA ) and ( $E$ )-3-ethoxycarbonyl-methylidene-6-methoxyphthalide (1lcA) (eluted second); the latter showed $\tau 1.06(\mathrm{~d}, J 8,4-\mathrm{H}), 2.63(\mathrm{~d}, J 2,7-\mathrm{H}), 2.66$ (dd, 88 and $2,5-\mathrm{H}$ ), and $3.99\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right)$. A very small trace of a fourth compound probably the $(Z)-6$-methoxyisomer ( 11 ddA ), was also detected [ $\left.\tau 4.26\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right)\right]$.

3-Ethoxycarbonylmethyl-5-methoxyphthalide (16A).-By the general procedure, hydrogenation of either $(E)$ - or $(Z)$-3-ethoxycarbonylmethylidene-5-methoxyphthalide [(12cA) or ( 12 dA ) ] gave the phthalide (16A), $\lambda_{\text {max. }} 284,255,247$, and 220 $\mathrm{nm} ; \vee_{\text {max. }} 1740$ and $1718 \mathrm{~cm}^{-1} ; \tau 2.18(\mathrm{~d}, J 8,7-\mathrm{H}), 2.95$ (dd, $J 8$ and $9,6-\mathrm{H}), 3.07(\mathrm{~d}, J 2,4-\mathrm{H}), 4.19\left(\mathrm{t}, J 7, \mathrm{CHCH}_{2}\right)$, 5.77 (q, J 7, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 6.11 (OMe), 7.13 (dd, J 7 and 2, $\mathrm{CHCH}_{2}$ ), and $8.72\left(\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ ) (Found: $M^{+}, 250.0836$. $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{O}_{5}$ requires $M, 250.0841$ ).

Reaction between 4-Dimethylaminophthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 4-dimethylaminophthalic anhydride ( 0.24 g ) and the phosphorane ( 0.44 g ) gave a mixture of mainly two isomers ( 0.28 g ) which were separated by silica-gel chromatography in petroleum-ether ( $1: 2$ ), to give the following: (i) (E)-5-dimethylamino-3-ethoxycarbonylmethylidenephthalide ( 12 cB ) (eluted first) which crystallised from ethanol as fine yellow needles, m.p. $138-139^{\circ} \mathrm{C}$; $\lambda_{\text {max. }} 398$, 307 infl ., and 291.5 nm ; $v_{\text {max. }} 1766,1699$, and $1645 \mathrm{~cm}^{-1}$; $\tau 1.80(\mathrm{~d}, J 2,4-\mathrm{H}), 2.43(\mathrm{~d}, J 8,7-\mathrm{H}), 3.29$ (dd, $J 8$ and 2 , $6-\mathrm{H}), 4.09(: \mathrm{CHCO} 2 \mathrm{Et}), 5.78\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.88\left(\mathrm{NMe}_{2}\right)$, and $8.67\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 261(100 \%)$, 216(29), 189(77), and $188(22)$ (Found: C, 64.5; H, 5.6. $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{NO}_{4}$ requires C., $64.4 ; \mathrm{H}, 5.7 \%$ ); and (ii) (Z)-5-dimethylamino-3-ethoxycarbonylmethylidenephthalide (12dB) (eluted second) which crystallised from methanol as yellow prisms, m.p. 162-163 ${ }^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 386,317.5$ infl., 305 infl., and 290 nm ; $v_{\text {max. }} 1784$, 1691 , and $1667 \mathrm{~cm}^{-1}$; $\tau 2.35(\mathrm{~d}, J 8,7-\mathrm{H}), 3.20$ (dd, $J 8$ and 2, 6-H), $3.30(\mathrm{~d}, J 2,4-\mathrm{H}), 4.26\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.74(\mathrm{q}$, $\left.J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) .6 .84\left(\mathrm{NMe}_{2}\right)$, and $8.65\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; $m / e 261(100 \%), 216(39), 189(77)$, and 188(31) (Found: C, 64.6 ; H, $\left.5.8{ }^{\circ}{ }_{6}^{\circ}\right)$.

Small traces ( $<5 \%$ ) of the ( $E$ )- and ( $Z$ )-6-dimethylaminoisomers $[(11 \mathrm{cB})$ and $(11 \mathrm{~dB})]$ were also detected $[\tau 2.32$ (d, $J 8,4-\mathrm{H})$ and $4.38\left(: \mathrm{C} H \mathrm{CO}_{2} \mathrm{Et}\right)$ respectively].

Hydrogenation of the Products from 4-Dimethylaminophthalic Anhydride.-By the general procedure, hydrogenation of either $(E)$ - or ( $Z$ )-5-dimethylamino-3-ethoxycar-
bonylmethylidenephthalide $[(12 \mathrm{cB})$ or ( 12 dB$)]$ gave 5 -di-methylamino-3-ethoxycarbonylmethylphthalide (16B) which crystallised from ether-cyclohexane as prisms, m.p. 67$68{ }^{\circ} \mathrm{C}, \lambda_{\text {max. }} 312$ and 298 infl. nm ; $\nu_{\text {max. }} 1752$ and $1740 \mathrm{~cm}^{-1}$; $\tau 2.45(\mathrm{~d}, J 8,7-\mathrm{H}), 3.37(\mathrm{dd}, J 8$ and $2,6-\mathrm{H}), 3.58(\mathrm{~d}, J 2$, $4-\mathrm{H}), 4.39\left(\mathrm{t}, J 7.5, \mathrm{CHCH}_{2}\right), 5.88\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.00$ $\left(\mathrm{NMe}_{2}\right), 7.24\left(\mathrm{~d}, J 7.5, \mathrm{CHCH}_{2}\right)$, and $8.77\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e$ $263(97 \%)$, 190(30), 189(59), 176(100), and 148(25) (Found: C, $64.1 ; \mathrm{H}, 6.5 . \quad \mathrm{C}_{14} \mathrm{H}_{17} \mathrm{NO}_{4}$ requires C, $63.9 ; \mathrm{H}, 6.5 \%$ ).

Reaction between 4-Methylphthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 4 -methylphthalic anhydride ( 1.62 g ) and the phosphorane $(3.48 \mathrm{~g})$ gave a mixture of all four possible isomers $(2.01 \mathrm{~g})$, chromatography of which over silica gel eluted with etherpetroleum ( $2: 3$ ) resulted in the complete resolution of the $(E)$-isomers from the $(Z)$ isomers but only gave a partial separation of the individual components. Crystallisation of one fraction from methanol gave a sample of pure (E)-3-ethoxycarbonylmethylidene-6-methylphthalide (11cC), as fine, colourless needles, m.p. $119-120^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 325.5,313.5,286$, 276, and 240 nm ; $\nu_{\text {max. }} 1790,1704$, and $1652 \mathrm{~cm}^{-1} ; \tau 1.05$ (d, $J 8,4-\mathrm{H}), 2.14(\mathrm{~d}, J 2,7-\mathrm{H}), 2.36$ (dd, $J 8$ and $2,5-\mathrm{H}$ ), $3.92\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.68\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.49\left(\mathrm{CH}_{3}\right)$, and 8.66 ( $\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $m / e 232(27 \%), 204(9), 187(100)$, and $160(58)$ (Found: C, 67.3; H, 5.2. $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{O}_{4}$ requires C, $67.3 ; \mathrm{H}$, $5.2 \%$ ). The remaining mixtures were not further separated.

Reaction between 4-Methoxycarbonylphthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 4-methoxycarbonylphthalic anhydride ( 1.03 g ) and the phosphorane ( 1.74 g ) gave a mixture of three isomers ( 1.2 g ) which were separated by chromatography over silica gel eluted with petroleum-ether ( $1: 1$ ) to give the following: (i) (E)-3-ethoxycarbonylmethylidene-6-methoxycarbonylphthalide (11cD) (eluted first) which crystallised from ethanol as prisms, m.p. $136-137{ }^{\circ} \mathrm{C}, \lambda_{\text {max }} 335,323,284$, 274, 254, and 231 nm ; $\nu_{\text {max. }} 1791,1720,1703$, and 1658 $\mathrm{cm}^{-1} ; \tau 0.88(\mathrm{~d}, J 8,4-\mathrm{H}), 1.42(\mathrm{~d}, J 2,7-\mathrm{H}), 1.57(\mathrm{dd}, J 8$, and 2, $5-\mathrm{H}$ ), $3.81\left(: \mathrm{CHCO} \mathrm{C}_{2} \mathrm{Et}\right), 5.71\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 6.05$ ( OMe ), and $8.70\left(\mathrm{t}, \mathrm{J} \mathrm{7}, \mathrm{CH}_{2} \mathrm{C} H_{3}\right) ; ~ m / e ~ 276(21 \%), ~ 248(16), ~$ 245(12), 232(16), 231(100), and 204(60) (Found: C, 61.1; $\mathrm{H}, 4.1 . \mathrm{C}_{14} \mathrm{H}_{12} \mathrm{O}_{6}$ requires $\mathrm{C}, 60.9$; $\mathrm{H}, 4.4 \%$ ); (ii) (E)-3-ethoxycarbonylmethylidene-5-methoxycarbonylphthalide (12cD) (eluted second), which crystallised from methanol as prisms, m.p. $119-120^{\circ} \mathrm{C}$, $\lambda_{\text {max }} 326.5,313.5,294,283,274 \mathrm{infl}$., and 232 nm ; $\nu_{\text {max }} 1796,1722,1705$, and $1660 \mathrm{~cm}^{-1} ; \tau 0.35(\mathrm{~d}$, $J 2,4-\mathrm{H}), 1.65(\mathrm{dd}, J 8$ and $2,6-\mathrm{H}), 1.97(\mathrm{~d}, J 8,7-\mathrm{H}), 3.83$ (: $\mathrm{CHCO}_{2} \mathrm{Et}$ ), 5.69 ( $\mathrm{q}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 6.03 ( OMe ), and 8.69 ( $\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) ; m/e $276(23 \%$ ), 248(18), 245(10), 232(16), $231(100)$, and $204(65)$ (Found: C, $60.9 ; \mathrm{H}, 4.2 \%$ ); and (iii) (Z)-3-ethoxycarbonylmethylidene-6-methoxycarbonylphthalide (11dD) (eluted last), which showed $\lambda_{\text {max. }} 324,310,292,283$, and 234 nm ; $\nu_{\text {max. }} 1791,1722,1705$, and $1660 \mathrm{~cm}^{-1}$; $\tau$ $1.38(\mathrm{~d}, J 2,7-\mathrm{H}), 1.55(\mathrm{dd}, J 8$, and $2,5-\mathrm{H}), 2.16(\mathrm{~d}, J 8$, $4-\mathrm{H}), 4.04\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.68\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 6.03(\mathrm{OMe})$, and $8.68\left(\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 276(70 \%), 232(20), 231(100)$, and 204(90) (Found: $M^{+}, 276.0639 . \mathrm{C}_{14} \mathrm{H}_{12} \mathrm{O}_{6}$ requires $M$, 276.0634).

Reaction between 4-Nitrophthalic Anhydride and the Phosphorane.-By the general procedure, reaction between 4 -nitrophthalic anhydride ( 0.193 g ) and the phosphorane $(0.348 \mathrm{~g})$ gave a mixture of three isomers which were separated by high-pressure liquid chromatography using chloroform-cyclohexane ( $1: 1$ ) as eluant to give the following: (i) (E)-3-ethoxycarbonylmethylidene-6-nitrophthalide (llcE) (eluted first) which crystallised from ethanol as
needles, m.p. $111-112^{\circ} \mathrm{C}, \lambda_{\text {max. }} 325,308,286,256$, and 224 $\mathrm{nm} ; \nu_{\text {max }} 1793,1710$, and $1660 \mathrm{~cm}^{-1} ; \tau 0.72(\mathrm{~d}, J 8,4-\mathrm{H})$, $1.28(\mathrm{~d}, J 2,7-\mathrm{H}), 1.38(\mathrm{dd}, J 8$ and $2,5-\mathrm{H}), 3.70\left(: \mathrm{CHCO}_{2}{ }^{-}\right.$ Et), $5.67\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, and $8.63\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e$ $263(3 \%)$, 234(32), 218(100), 191(94), and 172(91) (Found: C, $54.7 ; \mathrm{H}, 3.5 ; \mathrm{N}, 5.4 . \quad \mathrm{C}_{12} \mathrm{H}_{9} \mathrm{NO}_{6}$ requires $\mathrm{C}, 54.8 ; \mathrm{H}$, $3.4 ; \mathrm{N}, 5.3 \%$; (ii) (E)-3-ethoxycarbonylmethylidene-5-nitrophthalide (12cE) (eluted second) which crystallised from benzene-chloroform as prisms, m.p. $96-97{ }^{\circ} \mathrm{C}, \lambda_{\text {max. }} 327$, $310,290,255$, and 221 nm ; $v_{\max } 1791,1702$, and 1647 $\mathrm{cm}^{-1} ; \tau 0.11(\mathrm{~d}, J 2,4-\mathrm{H}), 1.47(\mathrm{dd}, J 8$, and $2,6-\mathrm{H}), 1.86$ $(\mathrm{d}, J 8,7-\mathrm{H}), 3.74(: \mathrm{CHCO} 2 \mathrm{Et}), 5.65\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, and 8.63 (t, J 7, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ); m/e 263(34\%), 234(15), 218(100), 191(47), and 172(65) (Found: C, 54.4; H, 3.6; N, 5.0\%); and (iii) (Z)-3-ethoxycarbonylmethylidene-6-nitrophthalide (lldE) (eluted last) which showed $\nu_{\max } 1790,1709$, and $1658 \mathrm{~cm}^{-1} ; ~ \tau 1.20(\mathrm{~d}, J 2,7-\mathrm{H}), 1.36(\mathrm{dd}, J 8$ and $2,5-\mathrm{H})$, $2.08(\mathrm{~d}, J 8,4-\mathrm{H}), 3.86\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.65\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, and 8.62 ( $\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) (Found: $M^{+}, 263.0419 . \mathrm{C}_{12}{ }^{-}$ $\mathrm{H}_{9} \mathrm{NO}_{6}$ requires $M, \mathbf{2 6 3 . 0 4 3 0}$ ).

Reaction between 4-Chlorophthalic Anhydride and the Phos-phorane.-By the general procedure, reaction between 4chlorophthalic anhydride ( 1.83 g ) and the phosphorane $(3.48 \mathrm{~g})$ gave a mixture of three isomers ( 2.0 g ). Chromatography over silica gel eluted with n-hexane-ether ( $9: 1$ ) gave (i) (E)-6-chloro-3-ethoxycarbonylmethylidenephthalide (11cF) (eluted first), which crystallised from methanol as soft plates, m.p. $111-113^{\circ} \mathrm{C}$; $\lambda_{\text {max. }} 325,313.5,285$ infl., and 275 nm ; $v_{\text {max }} 1788,1705$, and $1652 \mathrm{~cm}^{-1}$; $\tau 1.31(\mathrm{~d}, J 8,4-\mathrm{H})$, $2.38(\mathrm{~d}, J 2,7-\mathrm{H}), 2.55(\mathrm{dd}, J 8$ and $2,5-\mathrm{H}), 4.06\left(: \mathrm{CHCO}_{2}{ }^{-}\right.$ Et), $5.85\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, and 8.66 ( $\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ); $m / e 254(3 \%), 252(18), 226(2), 224(9), 209(31), 207(100)$, 182 (15), and 180 (52) (Found: C, 56.7; H, 3.5. $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{ClO}_{4}$ requires $\mathrm{C}, 57.0 ; \mathrm{H}, 3.6 \%$ ) ; (ii) (E)-5-chloro-3-ethoxycarbonylmethylidenephthalide ( 12 cF ) (eluted second), which crystallised from methanol as needles, m.p. $120-121^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 323,310,286 \mathrm{infl}$., and 278 nm ; $v_{\text {max }} 1794,1711$, and $1660 \mathrm{~cm}^{-1} ; \tau 1.25(\mathrm{~d}, J 2,4-\mathrm{H}), 2.38(\mathrm{~d}, J 8,7-\mathrm{H}), 2.61(\mathrm{dd}$, $J 8$ and $2,6-\mathrm{H}), 4.05\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.84\left(\mathrm{q}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, and $8.66\left(\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 254(4 \%)$, 252(19), 226(2), 224(10), 209(30), 207(100), 182(16), and 180(52) (Found: C, $57.0 ; \mathrm{H}, 3.4 \%$ ); and (iii) (Z)-5-chloro-3-ethoxycarbonylmethylidenephthalide (12dF) (eluted last) which crystallised from methanol as needles, m.p. $110-111{ }^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 321.5$, $310,284 \mathrm{infl}$., and 276 nm ; $\nu_{\text {max. }} 1804,1719$, and $1666 \mathrm{~cm}^{-1}$; $\tau 2.36(\mathrm{~d}, J 8,7-\mathrm{H}), 2.54(\mathrm{~d}, J 2,4-\mathrm{H}), 2.62(\mathrm{dd}, J 8$ and $2,6-\mathrm{H}), 4.31\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.83\left(\mathrm{q}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, and 8.68 ( $\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ); m/e 254(2\%), 252(5), 209(14), 207(100), 182(9), and $180(50)$ (Found: C, 57.4 ; H, 3.4\%).

Hydrogenation of the Products from 4-Chlorophthalic Anhydride.-By the general procedure, hydrogenation of (E)-6-chloro-3-ethoxycarbonylmethylidenephthalide (1lcF) $(21 \mathrm{mg})$ over $10 \% \mathrm{Pd}-\mathrm{C}$ gave the expected 6-chloro-3-ethoxycarbonylmethylphthalide ( 15 F ) ( 17 mg ) which recrystallised from n-hexane-ether (ca. $10: 1$ ) as clusters of thick, colourless needles, m.p. $95-96{ }^{\circ} \mathrm{C}$, $\lambda_{\text {max. }} 291,283$, and 278 infl. nm; $\nu_{\text {max. }} 1764$ and $1727 \mathrm{~cm}^{-1} ; \tau 2.18$ (d, $\left.J 2,7-\mathrm{H}\right), 2.40$ (dd, $J 8$, and $2,5-\mathrm{H}), 2.58(\mathrm{~d}, J 8,4-\mathrm{H}), 4.18\left(\mathrm{t}, J 7.5, \mathrm{C} H \mathrm{CH}_{2}\right)$, 5.82 (q, J 7, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 7.11 (bd, J 7.5, $\mathrm{CHCH}_{2}$ ), and 8.73 ( $\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) ; m/e 256(6\%), 254(17), 182(35), 180(100), $169(19), 167(73)$, and $146(22)$ (Found: C, 56.1; H, 4.0. $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{ClO}_{4}$ requires C , $56.6 ; \mathrm{H}, 4.3 \%$ ) (Found: $M^{+}$, 254.033. $\quad \mathrm{C}_{12} \mathrm{H}_{11}{ }^{35} \mathrm{ClO}_{4}$ requires $M, 254.035$ ).

Hydrogenation of either $(E)$ - or $(Z)$-5-chloro-3-ethoxycarbonylmethylidenephthalide $[(12 \mathrm{cF})$ or (12dF)] gave 5-
chloro-3-ethoxycarbonylmethylphthalide (16F) which crystallised from petroleum-ether as small, colourless needles, m.p. $105-106{ }^{\circ} \mathrm{C}, \lambda_{\text {max. }}$ 291.5infl., 283, and 275.5 $\mathrm{nm} ; v_{\text {max. }} 1766$ and $1730 \mathrm{~cm}^{-1} ; \tau 2.10(\mathrm{~d}, J 8,7-\mathrm{H}), 2.26$ (dd, $J 8$ max. $2,6-\mathrm{H}), 2.45(\mathrm{~d}, J 2,4-\mathrm{H}), 4.16(\mathrm{t}, J 7.5$, $\mathrm{CHCH}_{2}$ ), $5.78\left(\mathrm{q}, \mathrm{J} \mathrm{7}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.09\left(\mathrm{~d}, J 7.5, \mathrm{CHCH}_{2}\right)$, and 8.71 ( $\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ); m/e 256(8\%), 254(60), 182(36), $180(100), 169(16), 167(61)$, and $146(47)$ (Found: C, 56.4; H, $4.2 \%$ ).

Reaction between 4-Bromophthalic Anhydride and the Phosphorane.--By the general procedure, reaction between 4 -bromophthalic anhydride ( 0.23 g ) and the phosphorane $(0.35 \mathrm{~g})$ gave a mixture of two isomers $(0.24 \mathrm{~g})$ which were separated by h.p.l.c. using chloroform-petroleum (1:19) as eluant to give the following (i) (E)-6-bromo-3-ethoxycarbonylmethylidenephthalide (11cG) (eluted first) which crystallised from methanol as needles, m.p. $114-115{ }^{\circ} \mathrm{C}, \lambda_{\max .} 326,312$, 287 infl., 277 infl., 265 infl., $238 \mathrm{infl} .$, and $217 \mathrm{~nm} ; \vee_{\text {max. }} 1778$, 1700 , and $1642 \mathrm{~cm}^{-1} ; \tau 1.06(\mathrm{~d}, J 8,4-\mathrm{H}), 1.93(\mathrm{~d}, J 2$, $7-\mathrm{H}), 2.08(\mathrm{dd}, J 8$ and $2,5-\mathrm{H}), 3.86\left(\mathrm{CH}: \mathrm{CO}_{2} \mathrm{Et}\right), 5.70(\mathrm{q}$, $J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), and $8.64\left(\mathrm{t}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; m / e 298(25 \%)$, 296(25), 254(23), 253(100), 252(25), 251(100), 226(70), and $224(71)$ (Found: C, 48.3; H, 2.8. $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{BrO}_{4}$ requires C, $48.5 ; \mathrm{H}, 3.0 \%$ ) ; and (ii) (E)-5-bromo-3-ethoxycarbonylmethylidenephthalide ( 12 cG ), which crystallised from methanol as needles, m.p. $140-141{ }^{\circ} \mathrm{C}, \lambda_{\text {max }}$ 324, 311, 288infl., 277,253 infl., 234 , and 207 nm ; $v_{\text {max }} 1780,1698$, and 1644 $\mathrm{cm}^{-1}$; $\tau 0.78(4-\mathrm{H}), 2.20(6-$ and $7-\mathrm{H}), 3.86\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right), 5.68$ ( $\mathrm{q}, \mathrm{J} 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), and 8.64 ( $\mathrm{t}, J 7, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ) ; m/e 298(32\%), 296(32), 254(18), 253(100), 252(27), 251(100), 226(69), and $224(72)$ (Found: C, 48.4; H, 3.0\%). Traces of the two possible $(Z)$ isomers were also detected, $\tau 3.99\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right)$ and $4.03\left(: \mathrm{CHCO}_{2} \mathrm{Et}\right)$.

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## REFERENCES

${ }^{1}$ P. A. Chopard, R. F. Hudson, and R. J. G. Searle, Tetrahedron Lett., 1965, 2357.
${ }^{2}$ C. F. Ingham, R. A. Massy-Westropp, G. D. Reynolds, and W. D. Thorpe, Aust. J. Chem., 1975, 28, 2499; P. J. Babidge and R. A. Massy-Westropp, ibid., 1977, 30, 1629; R. A. MassyWestropp and M. F. Price, ibid., 1980, 33, 333.
${ }^{3}$ C. F. Ingham, R. A. Massy-Westropp, and G. D. Reynolds, Aust. J. Chem., 1974, 27, 1477; C. F. Ingham and R. A. MassyWestropp, ibid., p. 1491.
${ }^{4}$ D. W. Knight and G. Pattenden, J. Chem. Soc., Perkin Trans. 1, 1979, 62.
${ }^{5}$ D. R. Gedge and G. Pattenden, J. Chem. Soc., Perkin Trans. 1, 1979, 89 .
${ }_{6}$ P. A. Chopard, R. J. G. Searle, and F. H. Devitt, J. Org. Chem., 1965, 30, 1015.
${ }_{7}$ For a review, see W. Flitsch and S. R. Schindier, Synthesis, 1975, 685.
${ }^{8}$ D. W. Knight and C. D. Portas, Tetrahedron Lett., 1977, 4543. For an alternative synthesis, see B. M. Trost, G. T. Rivers, and J. M. Gold, J. Org. Chem., 1980, 45, 1835.
${ }^{9}$ See, for example, J. A. Elvidge and D. E. H. Jones, J. Chem. Soc. C, 1971, 2424.

10 R. L. N. Harris and J. L. Huppatz, Aust. J. Chem., 1977, 30, 2225; B. T. Brown, O. Johansen, G. F. Katekar, and W. H. F. Sasse, Pestic. Sci., 1973, 4, 473.
${ }_{11}$ A. Allahdad, M. J. Begley, and D. W. Knight, unpublished results.

12 See A. J. Speziale and D. E. Bissing, J. Am. Chem. Soc., 1963, 85, 3878; H. O. House, V. K. Jones, and G. A. Frank, J. Org. Chem., 1964, 29, 3327.
${ }^{13}$ M. Braun, Angew. Chem. Int. Ed. Engl., 1978, 17, 945; Tetrahedron Lett., 1979, 2885; 1980, 21, 3871.

14 J. S. Swenton, D. K. Jackson, M. J. Manning, and P. W. Raynolds, J. Am. Chem. Soc., 1978, 100, 6182; J. S. Swenton and P. W. Raynolds, ibid., p. 6188.
${ }^{15}$ D. G. Miller, S. Trenbeath, and C. J. Sih, Tetrahedron Lett., 1976, 1637.
${ }_{18}$ M. S. Newman and C. W. Muth, J. Am. Chem. Soc., 1950, 72, 5191; M. S. Newman and C. D. McCleary, ibid., 1941, 63, 1542.
${ }_{17}$ M. S. Newman and P. G. Scheurer, J. Am. Chem. Soc., 1956, 78, 5004 .

18 S. Gabriel, L. Kornfeld, and C. Grunert, Chem. Ber., 1924 57, 302. See also C. E. Castro, E. J. Gaughan, and D. C. Owsley, J. Org. Chem., 1966, 31, 4071.
${ }^{19}$ Cf. G. Pattenden and B. C. L. Weedon, J. Chem. Soc. C, 1968, 1984; 1997.

20 R. Grewe, Chem. Ber., 1938, $91,907$.
21 A. J. McAlees, R. McCrindle, and D. W. Sneddon, J. Chem. Soc., Perkin Trans. 1, 1977, 2030.
${ }_{22}$ H. N. Stephens, J. Am. Chem. Soc., 1921, 43, 1950.
${ }^{23}$ M. S. Newman and V. Lee, J. Org. Chem., 1977, $42,1478$.
${ }^{24}$ P. Fristch, Liebigs Ann. Chem., 1897, 296, 344.
25 E. H. Huntress and R. L. Shriner, Org. Synth., Coll. vol. 2, 1943, p. 459.

